

LETTER

Novel 2,4-Diarylaminopyrimidine Derivatives Containing Pyridine Moiety: Design, Synthesis, Crystal Structure and Biological Evaluation^①

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ABSTRACT A series of 2,4-diarylaminopyrimidine derivatives containing pyridine structure were designed and synthesized. The crystal structures of compounds **5d** and **5e** were obtained from X-ray diffraction. The crystal structure of **5d** (C₂₅H₂₀ClFN₆O₂) belongs to the monoclinic system, space group *P*2₁/*c* with *a* = 11.0500(10), *b* = 18.3045(17), *c* = 13.5646(9) Å and β = 122.806(5)°. **5e** (C₂₅H₁₉ClF₂N₆O₂) is of monoclinic system, space group *P*2₁/*c* with *a* = 10.9998(18), *b* = 18.517(3), *c* = 13.6355(16) Å and β = 123.315(9)°. The bioassay results showed all of the target compounds exhibited potential antiproliferative activities against MKN-45, HT-29, A549, K562 and GIST882 cell lines. Among them, compounds **5a**, **5c** and **5e** exhibited remarkable inhibitory activities against GIST882, K562 and A549 cell lines with IC₅₀ values of 0.68, 0.38 and 0.60 μM, respectively, which were comparable to that of the positive control foretinib.

Keywords: pyrimidine, pyridine, synthesis, X-ray diffraction, antitumor activity;

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1 INTRODUCTION

Cancer is one of the most health problems in the world. Although many classes of drugs were used for the treatment, the needs for safe and effective anticancer compounds are still significant target^[1, 2]. Pyridine and pyrimidine-based compounds have been reported to show remarkable antitumor activities by means of inhibiting multiple enzymes, and some of them are already being marketed or are under clinical/pre-clinical studies, such as imatinib, sorafenib, crizotinib, abiraterone, BMS-794833, altiratinib, and BMS-777607^[3-6]. Among above compounds, BMS-794833, altiratinib, and BMS-777607 are representative 4-phenoxy pyridine based type II c-Met kinase inhibitors^[7, 8]. Blocking c-Met kinase activity by small-molecule inhibitors has been identified as a

promising approach for the treatment of cancers^[9].

As described in Fig. 1, structurally, most of 4-phenoxy pyridine type II c-Met inhibitors may be disconnected into three moieties: a 4-phenoxy pyridine core (moiety A), a phenyl or substituted phenyl group (moiety B) and a linker bridge (moiety C). Moiety A and moiety B are crucial for kinase activity^[10-12]. With the goal of finding more 4-phenoxy pyridine-based antitumor agents, *N*-[4-(2-fluorophenoxy)pyridin-2-yl]cyclopropanecarboxamide was used as the moiety A. Pyrimidine-2,4-diamine was introduced into the moiety C *via* cyclization strategy. Furthermore, various substituents were introduced at the phenyl ring (moiety B) to investigate their effects on activities (Fig. 1). Accordingly, a series of 4-phenoxy pyridine derivatives were designed, synthesized and evaluated for their *in vitro* antiproliferative activities

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against MKN-45, HT-29, A549, K562 and GIST882 cancer cell lines.

2 EXPERIMENTAL

2.1 Materials and methods

Unless otherwise specified, all materials were obtained from commercial suppliers and were used without further purification. ^1H NMR spectra were recorded on a Bruker Biospin 600 MHz instrument using TMS as the internal standard. All chemical shifts were reported in ppm. IR spectra were recorded as KBr pellets on a Perkin-Elmer Spectrum one FT-IR spectrometer. MS spectra were obtained on an Agilent 6460 QQQ mass spectrometer (Agilent, USA) analysis system. Crystal data were obtained on a Bruker P4 X-diffractometer.

2.2 Synthesis of *N*-(4-chloropyridin-2-yl)-cyclopropylformamide (1)

Cyclopropanecarbonyl chloride (9.30 g, 89.00 mmol) was dissolved in dried CH_2Cl_2 (30 mL) and dropwise added to a mixture of 4-chloropyridin-2-amine (8.80 g, 68.45 mmol), Et_3N (20.78 g, 205.35 mmol) and CH_2Cl_2 (80 mL) in an ice bath, which was then removed to raise the temperature to room temperature and stirred for 12 h. The resulting mixture was sequentially washed with 20% K_2CO_3 (50 mL \times 3) and brine (50 mL \times 3), and the organic phase was separated, dried over anhydrous Na_2SO_4 , filtered, and the filtrate was evaporated under reduced pressure. The crude product obtained was purified by silica gel chromatography to give 9.78 g (yield: 73%) of **1** as a white solid. IR (KBr) cm^{-1} : 3242(-NH-), 1707(C=O), 1670, 1588(C=N), 1574(Ar), 1537(Ar), 1404, 1257, 1213, 1190, 1150, 960, 873, 824, 710; ^1H NMR (600 MHz, CDCl_3) δ 8.79 (s, 1H), 8.31 (s, 1H), 8.16 (d, $J = 5.4$ Hz, 1H), 7.03 (dd, $J = 5.4, 1.6$ Hz, 1H), 1.60~1.49 (m, 1H), 1.17~1.09 (m, 2H), 0.97~0.87 (m, 2H); MS (ESI) m/z (%): 197.1 $[\text{M}+\text{H}]^+$.

2.3 Synthesis of *N*-(4-(2-fluoro-4-nitro-phenoxy)pyridin-2-yl)cyclopropylformamide (2)

A stirring mixture of compound **8** (8.00 g, 40.68 mmol) and 2-fluoro-4-nitrophenol (15.98 g, 101.71 mmol) in chlorobenzene (100 mL) was refluxed for about 40 h. After cooling to room temperature, the reaction mixture was concentrated under reduced pressure to yield a pale solid. The solid was dissolved in CH_2Cl_2 (150 mL), washed with saturated K_2CO_3 aqueous solution (80 mL \times 4) and then brine (60 mL \times 4), dried over anhydrous Na_2SO_4 , and finally concentrated under reduced pressure to afford a brown solid, which was purified

by silica gel chromatography to give 7.13 g (yield: 55%) of **2** as a light yellow solid. IR (KBr) cm^{-1} : 3418(-NH-), 1687(C=O), 1617(C=N), 1529(Ar), 1493(Ar), 1354, 1300, 1272, 1180, 1070, 875, 798; ^1H NMR (600 MHz, $\text{DMSO}-d_6$) δ 11.00 (s, 1H), 8.43 (dd, $J = 10.3, 2.2$ Hz, 1H), 8.30 (d, $J = 5.7$ Hz, 1H), 8.19 (dd, $J = 9.0, 1.9$ Hz, 1H), 7.76 (d, $J = 2.2$ Hz, 1H), 7.61 (t, $J = 8.5$ Hz, 1H), 6.86 (dd, $J = 5.6, 2.2$ Hz, 1H), 2.04~1.95 (m, 1H), 0.78 (t, $J = 6.3$ Hz, 4H); MS (ESI) m/z (%): 318.1 $[\text{M}+\text{H}]^+$.

2.4 Synthesis of *N*-(4-(4-amino-2-fluorophenoxy)pyridin-2-yl)cyclopropanecarboxamide (3)

A mixture of compound **2** (6.00 g, 18.91 mmol), iron powder (5.28 g, 94.56 mmol), acetic acid (11.36 g, 189.10 mmol), water (20 mL) and ethyl acetate (100 mL) was heated to reflux for 2 h. After completion of the reaction as indicated by TLC, the mixture was filtered immediately. The organic layer of the filtrate was separated, washed with water, dried over anhydrous Na_2SO_4 and filtered, and the filtrate was evaporated under reduced pressure when white solid appeared, which was filtered to obtain 3.51 g (yield: 65%) of **3** as a white solid. IR (KBr) cm^{-1} : 3411(-NH-), 2026, 1737(C=O), 1618(C=N), 1510(Ar), 1427, 1207, 1163, 993, 956, 866, 822, 610; ^1H NMR (600 MHz, $\text{DMSO}-d_6$) δ 10.79 (s, 1H), 8.15 (d, $J = 5.7$ Hz, 1H), 7.59 (s, 1H), 6.95 (t, $J = 9.0$ Hz, 1H), 6.67~6.61 (m, 1H), 6.49 (dd, $J = 13.1, 2.2$ Hz, 1H), 6.40 (d, $J = 8.7$ Hz, 1H), 5.44 (s, 2H), 2.03~1.88 (m, 1H), 0.76 (br, 4H); MS (ESI) m/z (%): 288.1 $[\text{M}+\text{H}]^+$, 310.1 $[\text{M}+\text{Na}]^+$.

2.5 Synthesis of *N*-(4-(4-((2-chloropyrimidin-4-yl)amino)-2-fluorophenoxy)pyridin-2-yl)cyclopropane formamide (4)

A mixture of compound **3** (3.00 g, 10.44 mmol), 2,4-dichloropyrimidine (1.87 g, 12.53 mmol), and diisopropylethylamine (1.35 g, 10.44 mmol) in isopropanol (50 mL) was heated at reflux for 25 h. Upon cooling to room temperature, the mixture evaporated under reduced pressure. The residue was dissolved in dichloromethane (80 mL), and washed with water (50 mL \times 3). The organic layer was dried over anhydrous Na_2SO_4 and concentrated under reduced pressure to afford crude product, which was purified by silica gel chromatography to give 3.17 g (yield: 76%) of **4** as a white solid. ^1H NMR (600 MHz, $\text{DMSO}-d_6$) δ 10.90 (s, 1H), 10.34 (s, 1H), 8.25~8.22 (m, 2H), 7.88~7.60 (m, 2H), 7.50~7.30 (m, 2H), 6.85~6.71 (m, 2H), 2.03~1.92 (m, 1H), 0.84~0.72 (m, 4H); MS (ESI) m/z (%): 400.1 $[\text{M}+\text{H}]^+$, 422.0 $[\text{M}+\text{Na}]^+$.

2.6 General procedure for preparing the target compounds (5a~5f)

A mixture of compound **4** (0.20 g, 0.50 mmol), aromatic amine (1.3 equivalents) and *p*-toluenesulfonic acid (1.0 equivalent) in isopropanol (8.0 mL) was refluxed for about 18 h. After cooling to room temperature, the resultant precipitate was filtered and dried under vacuum to afford the target compounds **5a~5f** as white solid.

***N*-(4-(2-fluoro-4-((2-(phenylamino)pyrimidin-4-yl)amino)phenoxy)pyridin-2-yl)cyclopropanecarboxamide (5a)**

Yield: 67%; ¹H NMR (600 MHz, DMSO-*d*₆) δ 10.58 (d, *J* = 13.1 Hz, 1H), 9.37 (d, *J* = 11.5 Hz, 1H), 9.05~8.80 (m, 1H), 8.20~8.06 (m, 2H), 8.02 (d, *J* = 5.7 Hz, 1H), 7.81~7.62 (m, 3H), 7.37 (d, *J* = 8.4 Hz, 1H), 7.30~7.19 (m, 2H), 7.10 (t, *J* = 8.8 Hz, 1H), 6.91 (t, *J* = 7.2 Hz, 1H), 6.65~6.50 (m, 1H), 6.22 (d, *J* = 5.7 Hz, 1H), 2.04~1.89 (m, 1H), 0.91~0.83 (m, 2H), 0.79~0.72 (m, 2H); MS (ESI) *m/z*(%): 457.2 [M+H]⁺, 479.1 [M+Na]⁺.

***N*-(4-(2-fluoro-4-((2-(4-fluorophenyl)amino)pyrimidin-4-yl)amino)phenoxy)pyridin-2-yl)cyclopropanecarboxamide (5b)**

Yield: 59%; ¹H NMR (600 MHz, DMSO-*d*₆) δ 10.74~10.55 (m, 1H), 9.53~9.32 (m, 1H), 9.16~8.87 (m, 1H), 8.20~7.92 (m, 3H), 7.70 (d, *J* = 13.4 Hz, 3H), 7.41~7.28 (m, 1H), 7.19~7.04 (m, 1H), 7.04~6.88 (m, 2H), 6.64~6.50 (m, 1H), 6.31~6.12 (m, 1H), 5.86~5.82 (m, 1H), 2.03~1.88 (m, 1H), 0.96~0.58 (m, 4H); IR (KBr, cm⁻¹): 3258(-NH-), 2355, 1678(C=O), 1591, 1508(Ar), 1422(Ar), 1312, 1207, 982, 820; MS (ESI) *m/z* (%): 475.2 [M+H]⁺, 497.1 [M+Na]⁺.

***N*-(4-(4-((2-((4-chlorophenyl)amino)pyrimidin-4-yl)amino)-2-fluorophenoxy)pyridin-2-yl)cyclopropanecarboxamide (5c)**

Yield: 55%; ¹H NMR (600 MHz, DMSO-*d*₆) δ 10.67~10.45 (m, 1H), 9.50~9.31 (m, 1H), 9.18~8.92 (m, 1H), 8.26~7.96 (m, 3H), 7.79~7.67 (m, 3H), 7.35 (d, *J* = 7.3 Hz, 1H), 7.26~7.00 (m, 3H), 6.59 (dd, *J* = 5.7, 2.3 Hz, 1H), 6.24 (d, *J* = 5.7 Hz, 1H), 2.07~1.86 (m, 1H), 0.90~0.83 (m, 2H), 0.79~0.73 (m, 2H); IR (KBr, cm⁻¹): 3281(-NH-), 2353, 1672(C=O), 1585, 1416(Ar), 1308, 1204, 1094, 978, 804, 687; MS (ESI) *m/z* (%): 491.1 [M+H]⁺, 513.1 [M+Na]⁺.

***N*-(4-(4-((2-((3-chlorophenyl)amino)pyrimidin-4-yl)amino)-2-fluorophenoxy)pyridin-2-yl)cyclopropanecarboxamide (5d)**

Yield: 57%; ¹H NMR (600 MHz, DMSO-*d*₆) δ 10.51 (br, 1H), 9.36 (br, 1H), 9.17~9.04 (m, 1H), 8.13 (d, *J* = 5.7 Hz, 1H), 8.04 (d, *J* = 5.7 Hz, 1H), 7.95 (s, 1H), 7.92~7.83 (m, 1H), 7.74 (d, *J* = 1.9 Hz, 1H), 7.57 (d, *J* = 8.2 Hz, 1H), 7.41 (d, *J* = 8.8 Hz, 1H), 7.18 (t, *J* = 8.1 Hz, 1H), 7.12 (t, *J* = 8.8 Hz, 1H),

6.87 (d, *J* = 7.9 Hz, 1H), 6.60~6.55 (m, 1H), 6.31~6.22 (m, 1H), 2.03~1.84 (m, 1H), 0.91~0.86 (m, 2H), 0.79~0.74 (m, 2H); IR (KBr, cm⁻¹): 3305(-NH-), 2357, 1679(C=O), 1619, 1416, 1314, 1207, 1105, 990, 808; MS (ESI) *m/z* (%): 491.1 [M+H]⁺, 513.1 [M+Na]⁺.

***N*-(4-(4-((2-((3-chloro-4-fluorophenyl)amino)pyrimidin-4-yl)amino)-2-fluorophenoxy)pyridin-2-yl)cyclopropanecarboxamide (5e)**

Yield: 45%; ¹H NMR (600 MHz, DMSO-*d*₆) δ 10.76~10.50 (m, 1H), 9.48 (d, *J* = 13.0 Hz, 1H), 9.25 (d, *J* = 13.9 Hz, 1H), 8.17~8.10 (m, 1H), 8.06~7.97 (m, 3H), 7.71 (d, *J* = 3.4 Hz, 1H), 7.58 (br, 1H), 7.45~7.33 (m, 1H), 7.20~7.03 (m, 2H), 6.64~6.54 (m, 1H), 6.26 (t, *J* = 6.8 Hz, 1H), 1.97 (d, *J* = 4.1 Hz, 1H), 0.92~0.71 (m, 4H); IR (KBr, cm⁻¹): 3269(-NH-), 2349, 1678(C=O), 1591, 1413(Ar), 1310, 1259, 1207, 1101, 990, 808; MS (ESI) *m/z* (%): 509.1 [M+H]⁺, 531.1 [M+Na]⁺.

***N*-(4-(2-fluoro-4-((2-(*p*-tolylamino)pyrimidin-4-yl)amino)phenoxy)pyridin-2-yl)cyclopropanecarboxamide (5f)**

Yield: 62%; ¹H NMR (600 MHz, DMSO-*d*₆) δ 10.63 (s, 1H), 9.38 (s, 1H), 8.85 (s, 1H), 8.16~8.08 (m, 2H), 8.00 (d, *J* = 0.6 Hz, 1H), 7.72 (s, 1H), 7.57 (d, *J* = 8.3 Hz, 2H), 7.37 (d, *J* = 8.7 Hz, 1H), 7.10 (t, *J* = 8.9 Hz, 1H), 7.07~7.01 (m, 2H), 6.59 (dd, *J* = 5.7, 2.3 Hz, 1H), 6.19 (d, *J* = 5.7 Hz, 1H), 2.27 (s, 3H), 2.05~1.90 (m, 1H), 0.89~0.82 (m, 2H), 0.79~0.72 (m, 2H); IR (KBr, cm⁻¹): 3196(-NH-), 3017, 2363, 1680(C=O), 1586, 1524(Ar), 1420, 1308, 1206, 1101, 986, 804, 700; MS (ESI) *m/z* (%): 471.2 [M+H]⁺, 493.2 [M+Na]⁺.

2. 7 X-ray data collection and structure refinement

The white powder of compounds **5d** and **5e** was dissolved in ethanol/ethyl acetate/tetrahydrofuran = 5:2:3 (V/V/V) and 5:3:1 (V/V/V) mixed solvents, respectively. After slowly evaporating the solvents for several days, some single crystals suitable for X-ray analysis were obtained. The X-ray crystallography data for two crystals were collected on a Bruker APEX-II CCD automatic diffractometer with graphite-monochromatized MoK α radiation (λ = 0.71073 Å) using the ϕ and ω -scan mode at 296(2) K. The structure was solved by direct methods and refined with the SHELX crystallographic software package^[13] and expanded by Fourier technique. The non-hydrogen atoms were refined anisotropically. The hydrogen atoms bound to carbon were determined with theoretical calculations and those attached to nitrogen and oxygen were determined with successive difference Fourier syntheses. Empirical absorption correction was applied. The structure was solved by direct methods

using SADABS^[14]. The hydrogen atoms were placed at the calculated positions and refined as riding atoms with isotropic displacement parameters^[15]. Crystallographic data and experimental details of structural analyses for **5d** and **5e** are

summarized in Table 1. The hydrogen bonds of the title compound are listed in Table 2, and geometric parameters for it can be found in Table 3.

Table 1. Crystal Data of Compounds **5d** and **5e**

Compound	5d	5e
CCDC	2088950	2088951
Empirical formula	C ₂₅ H ₂₀ ClF ₆ N ₆ O ₂	C ₂₅ H ₁₉ ClF ₂ N ₆ O ₂
Formula weight	490.91	508.91
Crystal size (mm ³)	0.23 × 0.20 × 0.16	0.28 × 0.25 × 0.22
Crystal system	Monoclinic	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> (Å)	11.0500(10)	10.9998(18)
<i>b</i> (Å)	18.3045(17)	18.517(3)
<i>c</i> (Å)	13.5646(9)	13.6355(16)
α (°)	90	90
β (°)	122.806(5)	123.315(9)
γ (°)	90	90
Volume (Å ³)	2306.1(4)	2320.9(6)
<i>Z</i>	4	4
<i>D</i> _c (g·cm ⁻³)	1.411	1.456
Index ranges	-9 ≤ <i>h</i> ≤ 13 -21 ≤ <i>k</i> ≤ 21 -16 ≤ <i>l</i> ≤ 14	-12 ≤ <i>h</i> ≤ 13 -22 ≤ <i>k</i> ≤ 14 -16 ≤ <i>l</i> ≤ 15
Absorption coefficient (μ)	0.210	0.218
<i>F</i> (000)	1012	1048
Theta range for data collection (°)	2.104 to 24.998	2.099 to 24.998
Reflections collected	11625	11670
Independent reflections	4051	4072
<i>R</i> _{int}	0.0273	0.0437
Goodness-of-fit on <i>F</i> ²	1.060	1.054
<i>R</i> , <i>wR</i> (<i>I</i> > 2σ(<i>I</i>))	<i>R</i> = 0.0606, <i>wR</i> = 0.2068	<i>R</i> = 0.0495, <i>wR</i> = 0.1139
<i>R</i> , <i>wR</i> (all data)	<i>R</i> = 0.0866, <i>wR</i> = 0.2263	<i>R</i> = 0.0943, <i>wR</i> = 0.1282

$R = \sum(|F_o| - |F_c|) / \sum|F_o|$, $wR = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$

Table 2. Hydrogen Bond Lengths (Å) and Bond Angles (°) of **5d** and **5e**

	D—H⋯A	d(D—H)	d(H⋯A)	d(D⋯A)	∠DHA
5d	N(1)—H(1)⋯N(5)	0.86	2.26	3.084(4)	160
	N(4)—H(4)⋯O(2) ^{#1}	0.86	2.06	2.871(4)	157
5e	N(1)—H(1)⋯N(5) ^{#1}	0.86	2.24	3.051(3)	159
	N(4)—H(4)⋯O(2) ^{#2}	0.86	2.06	2.874(2)	157

Symmetry codes: **5d**: #1: -*x*, *y*+1/2, -*z*+1/2; **5e**: #1: -*x*, -*y*, -*z*; #2: *x*, -*y*-1/2, *z*-1/2

Table 3. Geometric Parameters of **5d** and **5e**

Bond	Dist. for 5d	Dist. for 5e	Angle	5d (°)	5e (°)
C(1)—N(1)	1.397(4)	1.409(3)	C(6)—C(1)—N(1)	124.3(3)	122.9(3)
C(1)—C(2)	1.403(5)	1.387(4)	C(6)—C(1)—C(2)	118.6(3)	119.2(3)
C(3)—Cl(1)	1.745(4)	1.728(3)	N(2)—C(7)—N(3)	127.8(3)	127.5(2)
C(7)—N(2)	1.334(4)	1.341(3)	N(3)—C(7)—N(1)	114.5(3)	114.8(2)
C(7)—N(1)	1.381(4)	1.364(3)	C(8)—C(9)—C(10)	116.6(3)	116.2(3)
C(8)—C(9)	1.365(5)	1.358(4)	C(12)—C(11)—N(4)	119.8(3)	121.1(2)
C(10)—N(4)	1.357(4)	1.361(3)	C(18)—C(17)—C(21)	120.4(3)	120.1(3)

To be continued

C(11)–C(12)	1.388(5)	1.396(4)	N(5)–C(19)–N(6)	114.7(3)	114.5(2)
C(11)–N(4)	1.401(4)	1.405(3)	C(22)–C(23)–C(24)	117.5(3)	115.9(3)
C(14)–O(1)	1.382(4)	1.398(3)	C(24)–C(23)–C(25)	58.5(3)	58.6(2)
C(17)–C(18)	1.377(4)	1.377(4)	C(7)–N(1)–C(1)	127.6(3)	127.6(2)
C(17)–O(1)	1.379(4)	1.379(3)	C(7)–N(2)–C(10)	116.0(3)	115.9(2)
C(19)–N(5)	1.333(4)	1.338(3)	C(7)–N(3)–C(8)	113.8(3)	113.9(2)
C(19)–N(6)	1.399(4)	1.398(3)	C(22)–N(6)–C(19)	126.2(3)	126.2(2)
C(22)–O(2)	1.226(4)	1.230(3)	C(17)–O(1)–C(14)	117.7(2)	116.8(2)
C(22)–N(6)	1.362(4)	1.358(3)	N(1)–C(1)–C(2)–C(3)	–177.3(3)	175.3(2)
C(22)–C(23)	1.471(5)	1.464(4)	O(1)–C(14)–C(15)–C(16)	178.5(3)	178.3(2)
C(23)–C(24)	1.507(5)	1.509(4)	N(2)–C(7)–N(1)–C(1)	–7.6(5)	6.1(4)
C(24)–C(25)	1.477(6)	1.469(4)	N(6)–C(22)–C(23)–C(25)	–133.3(3)	–163.5(2)

2.8 *In vitro* anticancer activity test of the target compounds

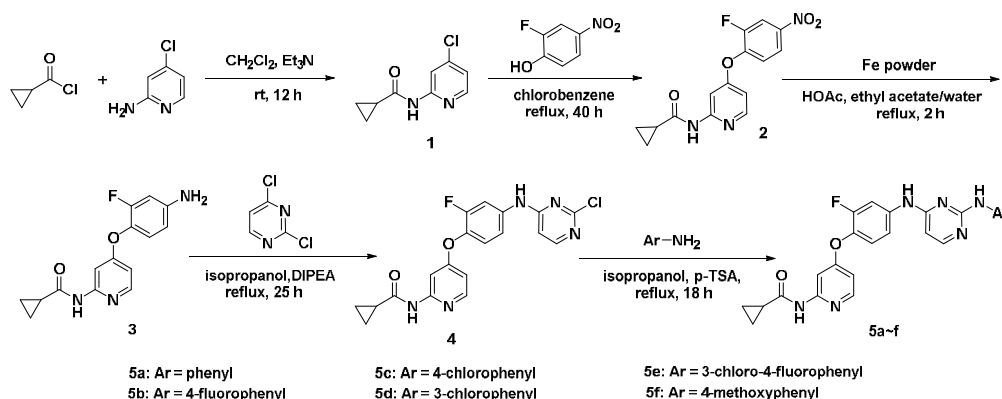
The antiproliferative activities of compounds **5a–5f** were evaluated against human gastric cancer cell lines MKN-45, human colon cancer cell lines HT-29, human lung adenocarcinoma cell lines A549, human chronic myeloid leukemia cell lines K562 and human gastrointestinal stromal tumor cell lines GIST882 using the standard MTT assay *in vitro*, with Foretinib as the positive control. The cancer cell lines were cultured in minimum essential medium (MEM) supplement with 10% fetal bovine serum (FBS). Approximate 4×10^3 cells, suspended in MEM medium, were plated onto each well of a 96-well plate and incubated in 5% CO₂ at 37 °C for 24 h. The tested compounds at the indicated final concentrations were added to the culture medium and the cell cultures were continued for 72 h. Fresh MTT was added to each well at a terminal concentration of 5 µg/mL, and incubated with cells at 37 °C for 4 h. The formazan crystals were dissolved in 100 mL DMSO each well, and the absorbency at 492 nm (for absorbance of MTT formazan) and 630 nm (for the reference wavelength) was measured with an ELISA reader. All compounds were tested three times in each

cell line. The results expressed as IC₅₀ (inhibitory concentration 50%) were the averages of three determinations and calculated by using the Bacus Laboratories Incorporated Slide Scanner (Bliss) software.

3 RESULTS AND DISCUSSION

3.1 Synthesis

The syntheses of target compounds **5a–5f** were outlined in Scheme 1. Commercially available 4-chloropyridin-2-amine was condensed with cyclopropanecarbonyl chloride in the presence of Et₃N to provide **1**, which underwent a nucleophilic substitution with 2-fluoro-4-nitrophenol to give the desired intermediate **2**. Reduction of the nitro group of **2** with iron powder and acetic acid in ethyl acetate/water (10:1 v/v) provided aniline compound **3**. **3** reacted with 2,4-dichloropyrimidine in refluxing *i*-PrOH to provide key intermediate **4** under the catalysis of DIPEA. **4** reacted with different substituted anilines in the presence of *p*-toluenesulfonic acid in refluxing *i*-PrOH to yield the target compounds **5a–5f**, and the their structures were confirmed by ¹H NMR, IR and ESI-MS.



Scheme 1. Synthetic route and structure of target compounds

3.2 Crystal structures

The structures of compounds **5d** and **5e** were further confirmed by single-crystal X-ray diffraction analysis. The molecular structures of compounds **5d** and **5e** with atom-numbering are shown in Fig. 2. Both crystal structures of **5d** and **5e** crystallize in monoclinic space group $P2_1/c$. Their crystal structures show that the two molecules have a 2,4-diarylaminopyrimidine skeleton, in which all bond lengths and bond angles fall in normal ranges. Both of the crystal structures consist of five rings: a three-membered cyclopropane (A), a pyrimidine ring (B), a pyridine ring (C) and two benzene rings (D, E). For compound **5d**, the molecule is not coplanar because the dihedral angles between rings A, B and C are 75.44° (A/C), 78.30° (A/B), 61.80° (A/D) and 33.91° (A/E), indicating rings B and C are almost perpendicular to

ring A. For compound **5e**, the molecule is also not coplanar because the dihedral angles between rings A, B, C, D and E are 75.12° (A/C), 77.42° (A/B), 61.87° (A/D) and 47.67° (A/E), so similar perpendicular conformation is also found. For compounds **5d** and **5e**, the bond lengths of C(22)–O(2) are 1.226(4) and 1.230(3) Å belonging to the typical C=O double bond^[16]. Meanwhile, hydrogen bonding interactions play a significant role in the crystal packing of **5d** and **5e** (Table 2). A mass of intermolecular hydrogen bonds N(4)–H(4A)⋯O(2), N(3)–H(6A)⋯N(6) and N(6)–H(6)⋯N(3) found in the two compounds play a major role in stabilizing the molecule. And the molecular structures of compounds **5d** and **5e** are shown in Fig. 3, depicting the molecular packing and hydrogen bonds in a unit cell. Within the molecule, the bond lengths and bond angles present no unusual features.

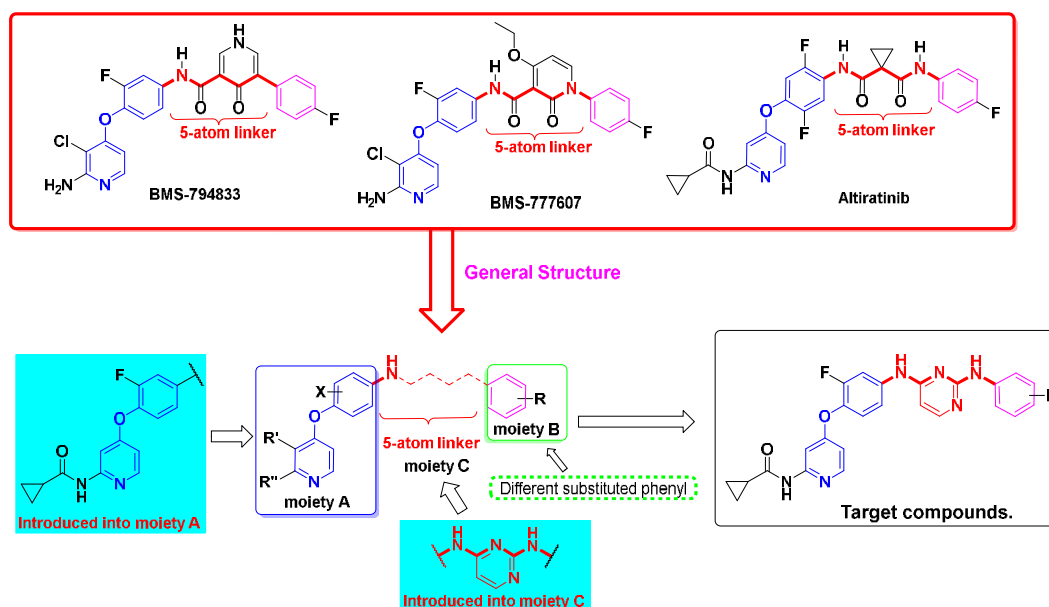


Fig. 1. The representative 4-phenoxy pyridine type II c-Met kinase inhibitors and design strategy of the target compounds

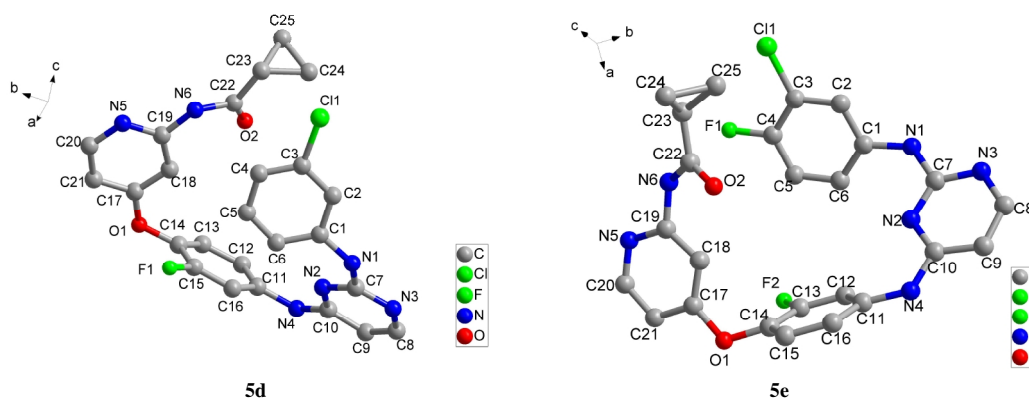
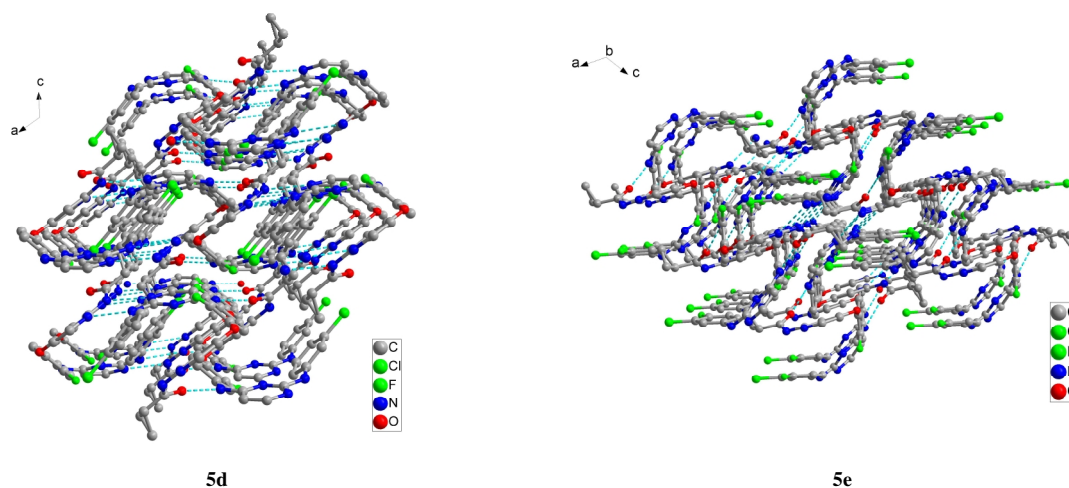


Fig. 2. Molecular structure of compound **5d** and **5e** in a single crystal at 296 K

Fig. 3. Crystal packing diagrams of compounds **5d** and **5e**

3.3 Biological activity analysis

The IC₅₀ values of **5a**–**5f** and Foretinib against MKN-45, HT-29, A549, K562 and GIST882 cancer cells are presented in Table 4, in which all the tested compounds show excellent antiproliferative activities against different cancer cells with IC₅₀ values ranging from 0.38 to 6.29 μM, indicating that newly synthesized 4-phenoxy pyridine derivatives maintained

antitumor activities. Notably, compounds **5a** (R₁ = H), **5c** (R₁ = 4-Cl) and **5e** (R₁ = 3-Cl-4-F) exhibited remarkable antiproliferative activities against GIST882, K562 and A549 cell lines with IC₅₀ values of 0.68 μM (foretinib: 0.75 μM), 0.38 μM (foretinib: 0.39 μM) and 0.60 μM (foretinib: 0.94 μM), respectively, which were all better than the positive control foretinib.

Table 4. Antiproliferative Activities of Target Compounds against Some Cancer Cells *in vitro*

Compd.	R ₁	IC ₅₀ (μM)				
		MKN-45	HT-29	A549	K562	GIST882
5a	H	2.38	3.51	2.34	0.95	0.68
5b	4-F	1.26	2.09	2.25	0.61	1.92
5c	4-Cl	0.62	1.05	1.50	0.38	1.54
5d	3-Cl	1.36	2.36	2.23	1.01	2.85
5e	3-Cl-4-F	0.75	2.92	0.60	0.65	2.44
5f	4-CH ₃	3.34	6.29	3.62	1.98	2.39
Foretinib	---	0.032	0.86	0.94	0.39	0.75

3.4 Physicochemical and ADME parameters

Furthermore, some physicochemical and ADME properties of the synthesized compounds and positive controls were predicted using the SwissADME (a free web tool to evaluate pharmacokinetics, drug likeness and medicinal chemistry friendliness of small molecules) for their adaptability with Lipinski's rule of five^[17-19]. Compounds obeying at least three of the four criteria are considered to adhere to Lipinski Rule.

As demonstrated in Table 5, the most active compounds show variable permeability based on gastrointestinal absorption (GI), according to the BOILED-Egg predictive

model (Brain Or IntestinaL EstimateD permeation method). All predicted compounds showed high gastrointestinal absorption. With respect to oral bioavailability, it is expected 0.55 of probability of oral bioavailability score >10% in the rat for all compounds, the same as the control drug foretinib (0.55). Compound **5a**–**5f** exhibited potent *in vitro* antitumor activity, low toxicity and reasonable physicochemical properties, because suitable flexible and size in the bioavailability radar map (Fig. 4). All these data suggests that compound **5a**–**5f** could be considered as a candidate for further research.

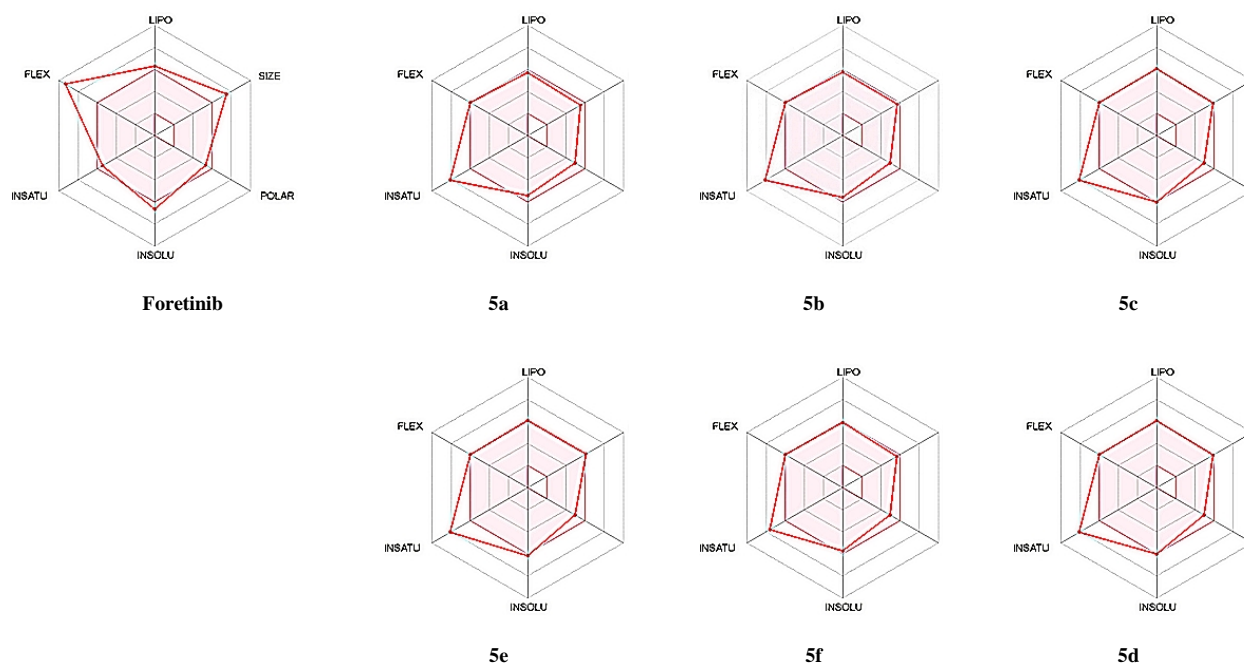


Fig. 4. The bioavailability radar enables a first glance at the drug-likeness of target compounds and Foretinib. The pink area represents the optimal range for each properties (lipophilicity: XLOGP3 between -0.7 and $+5.0$, size: MW between 150 and 500 g/mol, polarity: TPSA between 20 and 130 Å², solubility: logS not higher than 6 , saturation: fraction of carbons in the sp^3 hybridization not less than 0.25 , and flexibility: no more than 9 rotatable bonds).

Table 5. Physicochemical Properties and ADME Properties of Target Compounds

Compd.	MW (g/mol)	H-bond acceptors	H-bond donors	Log P o/w	Violation Lipinski Rule of 5	^a PSA (Å ²)	Rotatable bonds	^b BBB	^c GI	^d BS
	<500	<10	<5	<5		≤140	<10			
5a	456.47	6	3	3.94	0	101.06	9	NO	Low	0.55
5b	474.46	7	3	4.13	0	101.06	9	NO	Low	0.55
5c	490.92	6	3	4.46	0	101.06	9	NO	Low	0.55
5d	490.92	6	3	4.41	0	101.06	9	NO	Low	0.55
5e	508.91	7	3	4.66	1	101.06	9	NO	Low	0.55
5f	470.50	6	3	4.12	0	101.06	9	NO	Low	0.55
Foretinib	632.65	10	2	4.80	1	111.25	14	NO	Low	0.55

^aPSA – polar surface area. ^bBBB – blood-brain barrier. ^cGI – gastrointestinal absorption. ^dBS – bioavailability score.

Electronic supplementary information

Supplementary data associated with this article have been deposited in the ESI.

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