

Tuning Solvent Composition to Enhance the Stability of Metal Clusters in Mass Spectrometry

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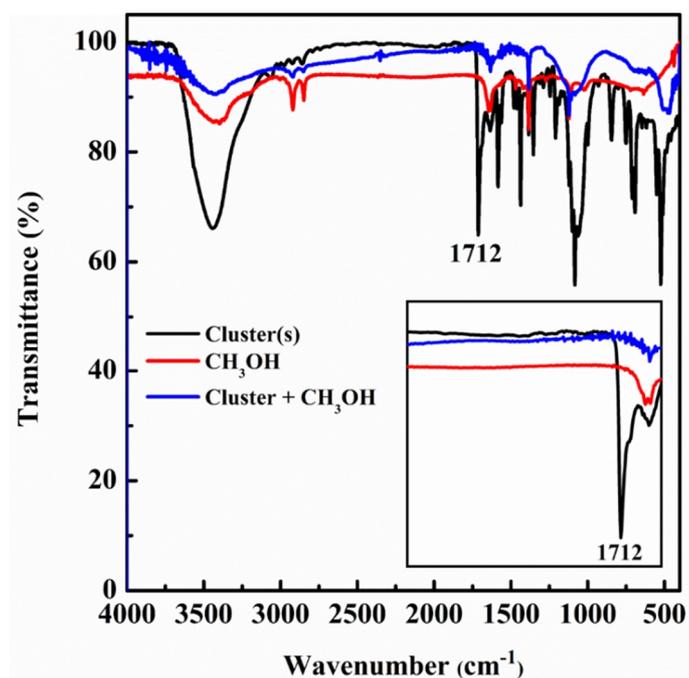


Figure S1. IR spectra of the cluster powders (black trace), CH₃OH (red trace), the cluster dissolved in CH₃OH (blue trace), respectively. Inset: the enlarged details in the range of 2300 to 1600 cm⁻¹. The C=O vibration peak at 1712 cm⁻¹ almost disappears when CH₃OH is applied, implying the occurrence of hemiacetal reaction between the cluster ligands and solvent molecules.

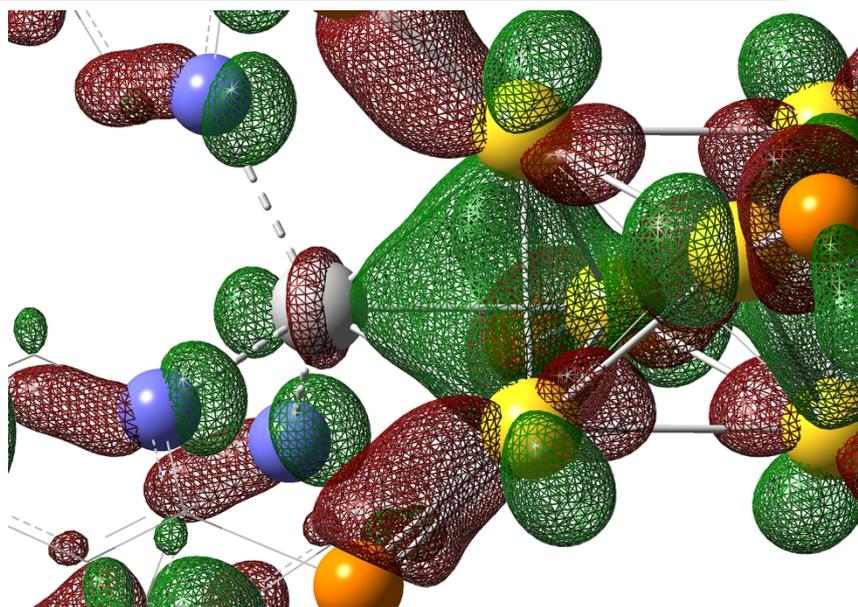


Figure S2. Enlarged part of Figure 4c illustrating the bonding properties between Ag and Au.

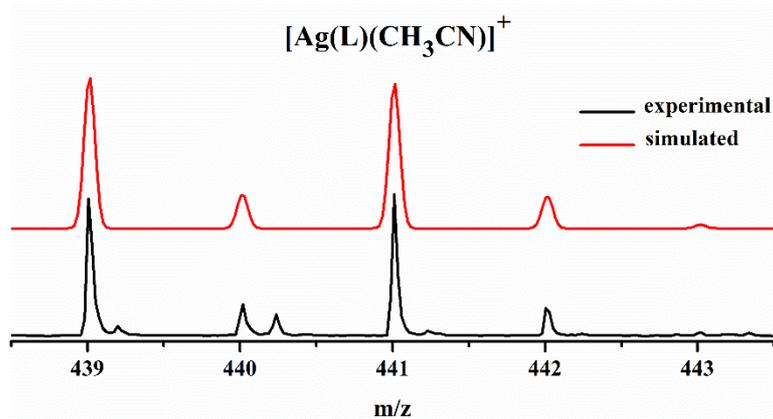


Figure S3. The experimental (black trace) and simulated (red trace) isotopic patterns of the fragment ion peak $[\text{Ag(L)(CH}_3\text{CN)}]^+$ obtained by HR-ESI-TOF-MS when the cluster dissolved in CH_3CN .